## organic compounds



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### **L-Leucylglycylglycine**

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Key indicators: single-crystal X-ray study; T = 173 K; mean  $\sigma(C-C) = 0.003$  Å; R factor = 0.038; wR factor = 0.087; data-to-parameter ratio = 16.4.

In the title compound,  $C_{10}H_{19}N_3O_4$ , the N- and C-termini are protonated and ionized, respectively, and the molecule forms a zwitterion. The main chain is in a folded form. In the crystal, the N-terminal  $-NH_3^+$  group hydrogen bonds to three C-terminal -COO groups and one carbonyl O atom, forming a three-dimensional network. In addition, an  $N-H\cdots O$  hydrogen bond between the amide groups of the middle glycine residue and a  $C-H\cdots O$  interaction continue along the a-axis direction. The side chains of the leucyl residues form a hydrophobic region along the a axis.

#### **Related literature**

For related structures of L-leucylglycylglycine, see: Goswami *et al.* (1977); Srikrishnan & Parthasarathy (1987); Kiyotani & Sugawara (2012).

#### **Experimental**

Crystal data

 $C_{10}H_{19}N_3O_4$  $M_r = 245.28$  Orthorhombic,  $P2_12_12_1$ a = 5.391 (5) Å b = 11.742 (10) Å c = 19.975 (16) Å  $V = 1264.4 (19) \text{ Å}^3$ Z = 4 Mo  $K\alpha$  radiation  $\mu = 0.10 \text{ mm}^{-1}$  T = 173 K $0.48 \times 0.18 \times 0.08 \text{ mm}$ 

Data collection

Rigaku Mercury CCD area-detecter diffractometer 9374 measured reflections 2887 independent reflections 2200 reflections with  $I > 2\sigma(I)$  $R_{\text{int}} = 0.070$ 

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.038$   $wR(F^2) = 0.087$  S = 0.962887 reflections 176 parameters 3 restraints H atoms treated by a mixture of independent and constrained refinement

 $\Delta \rho_{\text{max}} = 0.14 \text{ e Å}^{-3}$  $\Delta \rho_{\text{min}} = -0.17 \text{ e Å}^{-3}$ 

## **Table 1** Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D-\mathbf{H}\cdot\cdot\cdot A$
N1-H1···O3i	0.97 (3)	1.78 (3)	2.743 (2)	168 (2)
$N1-H2\cdots O4^{ii}$	0.92(2)	1.94(2)	2.822 (2)	160 (2)
$N1-H2\cdots O3^{ii}$	0.92(2)	2.52 (2)	3.260 (2)	138 (2)
N1-H3···O1 <sup>iii</sup>	0.89(2)	2.45 (2)	3.031(2)	123 (2)
$N1-H3\cdots O3^{iv}$	0.89(2)	2.05(2)	2.870(2)	151 (2)
$N2-H5\cdots O2^{v}$	0.83 (2)	2.05 (2)	2.832 (2)	155 (2)
$C1-H4\cdots O1^{v}$	1.00	2.33	3.269 (2)	155

Symmetry codes: (i)  $-x+\frac{1}{2},-y+1,z-\frac{1}{2};$  (ii)  $-x-\frac{1}{2},-y+1,z-\frac{1}{2};$  (iii)  $x-\frac{1}{2},-y+\frac{3}{2},-z+1;$  (iv)  $-x,y+\frac{1}{2},-z+\frac{3}{2};$  (v) x-1,y,z.

Data collection: CrystalClear (Rigaku, 2006); cell refinement: CrystalClear; data reduction: CrystalClear; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: Mercury (Macrae et al., 2006); software used to prepare material for publication: SHELXL97 (Sheldrick, 2008) and Yadokari-XG 2009 (Kabuto et al., 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: IS5257).

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## supplementary materials

Acta Cryst. (2013). E69, o660 [doi:10.1107/S1600536813008490]

## L-Leucylglycylglycine

#### Masanori Ootaki, Yukino Nawa, Tomoko Hiroi, Hiroaki Matsui and Yoko Sugawara

#### Comment

A moderately large number of oligopeptides are biologically active, and their structures are investigated to facilitate the determination of the possible conformations of oligopeptide and polypeptide chains. The N-terminus and C-terminus of *L*-leucylglycylglycine (*L*-LGG) are protonated and ionized, respectively, and the molecule is a zwitterion (Fig. 1). The main chain is in a folded form. The torsion angles of C2—N2—C3—C4 and N2—C3—C4—N3 are -87.54 (18)° and -52.78 (19)°, respectively. The N-terminal –NH<sub>3</sub>+ groups and the C-terminal –COO- groups form hydrogen bond networks (Fig. 2 & Fig. 3). One of the hydrogen atoms of the –NH<sub>3</sub> group forms a three-centered hydrogen bond with the carboxyl and carboxyl oxygen atoms. In addition, intermolecular hydrogen bonds among amide groups (NH···O=C) are formed along the *a* axis. The hydrophobic region composed of leucyl side chains is surrounded by hydrophilic parts, and forms a column along the *a* axis.

In the case of the L-leucylglycylglycylglycine (Srikrishnan & Parthasarathy, 1987), the main chain is a folded form, and hydrophobic columns are formed along the a axis as in the case of L-LGG. On the other hand, the main chain of D, L-leucylglycylglycine (D, L-LGG) (Goswami et al., 1977) is in a nearly all-trans form expect the N-terminus. The main chains align parallel to the b axis in a head-to-tail manner and a  $\beta$ -sheetlike structure is formed parallel to the bc plane. The hydrophobic regions of the leucyl side chains and the hydrophilic regions are aligned alternately along the a axis. As in the case of D, L-LGG, the main chain in L-leucylglycine 0.67 hydrate (Kiyotani & Sugawara, 2012) is in a extended form, and hydrophobic and hydrophilic regions are aligned alternately along the c axis.

#### **Experimental**

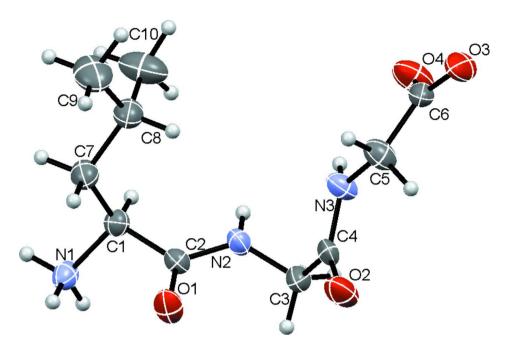
L-Leucylglycylglycine was purchased from Bachem Inc. Single crystals were obtained from an aqueous solution.

#### Refinement

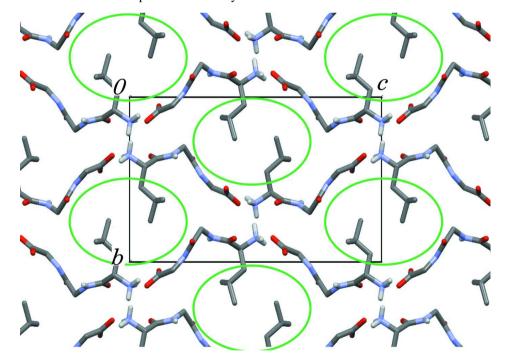
H atoms were placed in calculated positions with C—H = 0.98 Å (CH<sub>3</sub>), 0.99 Å (CH<sub>2</sub>) or 1.00 Å (CH) and refined in a riding mode with  $U_{iso}(H) = 1.2 U_{eq}(C)$ . The other H atoms were placed in a difference Fourier map. The N-terminal H atoms were restrained to N—H = 0.87 (4) Å during refinements. The absolute configuration was known for the purchased material.

#### **Computing details**

Data collection: *CrystalClear* (Rigaku, 2006); cell refinement: *CrystalClear* (Rigaku, 2006); data reduction: *CrystalClear* (Rigaku, 2006); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: Mercury (Macrae *et al.*, 2006); software used to prepare material for publication: *SHELXL97* (Sheldrick, 2008) and *Yadokari-XG* 2009 (Kabuto *et al.*, 2009).



**Figure 1**View of the title compound with the atomic numbering scheme. Displacement ellipsoids are drawn at 50% probability level and H atoms are shown as small spheres of arbitrary radii.



**Figure 2** Packing diagram of the title compound viewed along the *a* axis. Hydrophobic columns are indicated by green circles.

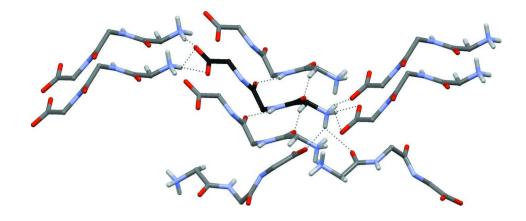


Figure 3

Hydrogen bonding scheme around the molecule, whose carbon atoms are colored with black. Hydrogen bonds are indicated by dotted lines. Side-chain atoms of the leucyl residues have been omitted for clarity.

#### L-Leucylglycylglycine

Crystal data

 $C_{10}H_{19}N_3O_4$   $M_r = 245.28$ Orthorhombic,  $P2_12_12_1$ Hall symbol: P 2ac 2ab a = 5.391 (5) Å b = 11.742 (10) Å c = 19.975 (16) Å V = 1264.4 (19) Å<sup>3</sup> Z = 4

Data collection

Rigaku Mercury CCD area-detecter diffractometer Radiation source: rotating anode Graphite monochromator ω scans 9374 measured reflections 2887 independent reflections

Refinement

Refinement on  $F^2$ Least-squares matrix: full  $R[F^2 > 2\sigma(F^2)] = 0.038$   $wR(F^2) = 0.087$  S = 0.962887 reflections 176 parameters 3 restraints Primary atom site location: structure-invariant direct methods F(000) = 528  $D_{\rm x} = 1.288~{
m Mg~m^{-3}}$ Mo Klpha radiation,  $\lambda = 0.71070~{
m Å}$ Cell parameters from 949 reflections  $\theta = 7.6{-}17.5^{\circ}$   $\mu = 0.10~{
m mm^{-1}}$   $T = 173~{
m K}$ Block Rod, colorless  $0.48 \times 0.18 \times 0.08~{
m mm}$ 

2200 reflections with  $I > 2\sigma(I)$   $R_{\rm int} = 0.070$   $\theta_{\rm max} = 27.5^{\circ}, \ \theta_{\rm min} = 3.5^{\circ}$   $h = -6 \rightarrow 6$   $k = -15 \rightarrow 15$   $l = -21 \rightarrow 25$ 

Secondary atom site location: difference Fourier map Hydrogen site location: inferred from neighbouring sites H atoms treated by a mixture of independent and constrained refinement  $w = 1/[\sigma^2(F_o^2) + (0.0374P)^2]$  where  $P = (F_o^2 + 2F_c^2)/3$   $(\Delta/\sigma)_{\rm max} = 0.012$   $\Delta\rho_{\rm max} = 0.14 \ {\rm e} \ {\rm Å}^{-3}$   $\Delta\rho_{\rm min} = -0.17 \ {\rm e} \ {\rm Å}^{-3}$ 

#### Special details

**Geometry**. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted R-factor wR and goodness of fit S are based on  $F^2$ , conventional R-factors R are based on F, with F set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on  $F^2$  are statistically about twice as large as those based on F, and F-factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters  $(\mathring{A}^2)$ 

	x	У	Z	$U_{ m iso}$ * $/U_{ m eq}$	
N1	-0.1744 (3)	0.64554 (12)	0.49670 (7)	0.0348 (3)	
N2	-0.0484(3)	0.64749 (12)	0.67001 (6)	0.0315 (3)	
H5	-0.195(4)	0.6402 (15)	0.6825 (9)	0.036 (5)*	
N3	0.1814(3)	0.52895 (12)	0.78349 (7)	0.0362 (3)	
H8	0.022(4)	0.5354 (16)	0.7880 (9)	0.048 (6)*	
O1	0.2239 (2)	0.64148 (11)	0.58454 (5)	0.0415 (3)	
O2	0.5246(2)	0.62376 (11)	0.75146 (6)	0.0485 (3)	
C1	-0.1911(3)	0.58011 (12)	0.56075 (8)	0.0290(3)	
H4	-0.3570	0.5921	0.5820	0.035*	
C2	0.0137(3)	0.62595 (12)	0.60611 (7)	0.0273 (3)	
C3	0.1320(3)	0.69772 (14)	0.71523 (8)	0.0368 (4)	
Н6	0.0422	0.7431	0.7493	0.044*	
H7	0.2381	0.7507	0.6894	0.044*	
C4	0.2975 (3)	0.61277 (13)	0.75092 (7)	0.0314 (4)	
C5	0.3059(3)	0.45098 (16)	0.82883 (9)	0.0406 (4)	
H9	0.4550	0.4883	0.8477	0.049*	
H10	0.3604	0.3826	0.8038	0.049*	
C6	0.1333 (4)	0.41541 (13)	0.88559 (8)	0.0354 (4)	
O3	0.2277 (3)	0.35522 (11)	0.93090 (6)	0.0536 (4)	
O4	-0.0855(3)	0.44525 (12)	0.88306 (7)	0.0545 (4)	
C7	-0.1526(3)	0.45349 (13)	0.54488 (8)	0.0374 (4)	
H11	-0.2711	0.4312	0.5094	0.045*	
H12	0.0168	0.4434	0.5267	0.045*	
C8	-0.1855(4)	0.37232 (14)	0.60441 (9)	0.0413 (4)	
H13	-0.0679	0.3959	0.6406	0.050*	
C9	-0.1199(5)	0.25116 (15)	0.58278 (10)	0.0636 (7)	
H14	-0.1379	0.1995	0.6210	0.076*	
H15	0.0518	0.2492	0.5667	0.076*	
H16	-0.2316	0.2272	0.5467	0.095*	
C10	-0.4475(5)	0.3769 (2)	0.63228 (13)	0.0715 (7)	
H17	-0.5664	0.3609	0.5964	0.086*	
H18	-0.4796	0.4529	0.6507	0.086*	
H19	-0.4656	0.3199	0.6678	0.086*	
H1	-0.011(5)	0.636(2)	0.4768 (12)	0.089 (9)*	
H2	-0.284 (5)	0.6147 (19)	0.4666 (12)	0.076 (7)*	
Н3	-0.202 (4)	0.7193 (16)	0.5047 (11)	0.058 (6)*	

# supplementary materials

## Atomic displacement parameters $(\mathring{A}^2)$

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
N1	0.0420 (9)	0.0301 (7)	0.0322 (7)	-0.0012 (7)	-0.0133 (7)	0.0028 (6)
N2	0.0312 (8)	0.0404 (8)	0.0228 (6)	-0.0052 (7)	0.0020(6)	0.0025 (6)
N3	0.0288 (8)	0.0458 (8)	0.0341 (7)	-0.0020(7)	0.0000(6)	0.0152 (6)
O1	0.0248 (6)	0.0666 (8)	0.0332 (6)	-0.0019(6)	0.0030 (5)	0.0004 (6)
O2	0.0387 (8)	0.0668 (9)	0.0399 (7)	-0.0099(7)	0.0061 (5)	0.0128 (6)
C1	0.0284 (8)	0.0304 (7)	0.0282 (8)	0.0006 (7)	-0.0021(7)	0.0001 (6)
C2	0.0272 (8)	0.0294(8)	0.0253 (7)	0.0018 (7)	0.0018 (6)	0.0038 (6)
C3	0.0496 (12)	0.0340 (8)	0.0268 (8)	-0.0083 (8)	-0.0054(8)	0.0034 (7)
C4	0.0375 (10)	0.0396 (9)	0.0172 (7)	-0.0051 (8)	0.0031 (7)	0.0006 (6)
C5	0.0365 (10)	0.0480 (10)	0.0373 (9)	0.0046 (9)	0.0010(8)	0.0148 (8)
C6	0.0498 (11)	0.0301(8)	0.0264(8)	-0.0120 (8)	-0.0017(8)	0.0028 (7)
O3	0.0698 (9)	0.0532 (7)	0.0379 (7)	-0.0196 (7)	-0.0195 (7)	0.0220(6)
O4	0.0427 (8)	0.0644 (9)	0.0562 (8)	0.0010(7)	0.0156 (7)	0.0178 (7)
C7	0.0482 (11)	0.0303 (8)	0.0338 (9)	0.0007 (8)	-0.0012(8)	-0.0004(7)
C8	0.0532 (11)	0.0325 (8)	0.0381 (9)	-0.0025 (9)	-0.0036(8)	0.0037 (8)
C9	0.096(2)	0.0342 (10)	0.0602 (14)	0.0078 (10)	0.0029 (13)	0.0088 (9)
C10	0.0713 (16)	0.0552 (13)	0.0882 (17)	-0.0017 (12)	0.0223 (13)	0.0234 (12)

## Geometric parameters (Å, °)

N1—C1	1.495 (2)	C5—C6	1.525 (3)
N1—H1	0.97(2)	C5—H9	0.9900
N1—H2	0.92(2)	C5—H10	0.9900
N1—H3	0.894 (19)	C6—O4	1.232 (3)
N2—C2	1.344 (2)	C6—O3	1.256 (2)
N2—C3	1.452 (2)	C7—C8	1.534 (2)
N2—H5	0.83(2)	C7—H11	0.9900
N3—C4	1.336 (2)	C7—H12	0.9900
N3—C5	1.452 (2)	C8—C10	1.519 (3)
N3—H8	0.87(2)	C8—C9	1.528 (3)
O1—C2	1.226(2)	C8—H13	1.0000
O2—C4	1.231 (2)	C9—H14	0.9800
C1—C2	1.526 (2)	C9—H15	0.9800
C1—C7	1.534(2)	C9—H16	0.9800
C1—H4	1.0000	C10—H17	0.9800
C3—C4	1.516 (2)	C10—H18	0.9800
C3—H6	0.9900	C10—H19	0.9800
C3—H7	0.9900		
C1—N1—H1	110.2 (15)	C6—C5—H9	109.5
C1—N1—H2	108.6 (14)	N3—C5—H10	109.5
H1—N1—H2	105 (2)	C6—C5—H10	109.5
C1—N1—H3	109.6 (14)	H9—C5—H10	108.1
H1—N1—H3	110 (2)	O4—C6—O3	125.31 (17)
H2—N1—H3	113 (2)	O4—C6—C5	118.43 (16)
C2—N2—C3	120.03 (15)	O3—C6—C5	116.25 (18)
C2—N2—H5	120.03 (13)	C8—C7—C1	115.23 (14)
	` ′		` ′

# supplementary materials

C3—N2—H5	119.4 (13)	C8—C7—H11	108.5
C4—N3—C5	123.47 (17)	C1—C7—H11	108.5
C4—N3—H8	116.8 (13)	C8—C7—H12	108.5
C5—N3—H8	116.7 (13)	C1—C7—H12	108.5
N1—C1—C2	106.45 (13)	H11—C7—H12	107.5
N1—C1—C7	108.24 (13)	C10—C8—C9	110.59 (17)
C2—C1—C7	111.49 (13)	C10—C8—C7	111.68 (16)
N1—C1—H4	110.2	C9—C8—C7	109.41 (15)
C2—C1—H4	110.2	C10—C8—H13	108.4
C7—C1—H4	110.2	C9—C8—H13	108.4
O1—C2—N2	122.41 (14)	C7—C8—H13	108.4
O1—C2—C1	120.84 (14)	C8—C9—H14	109.5
N2—C2—C1	116.75 (14)	C8—C9—H15	109.5
N2—C3—C4	114.79 (14)	H14—C9—H15	109.5
N2—C3—H6	108.6	C8—C9—H16	109.5
C4—C3—H6	108.6	H14—C9—H16	109.5
N2—C3—H7	108.6	H15—C9—H16	109.5
C4—C3—H7	108.6	C8—C10—H17	109.5
H6—C3—H7	107.5	C8—C10—H18	109.5
O2—C4—N3	122.62 (16)	H17—C10—H18	109.5
O2—C4—C3	121.36 (15)	C8—C10—H19	109.5
N3—C4—C3	115.97 (17)	H17—C10—H19	109.5
N3—C5—C6	110.74 (16)	H18—C10—H19	109.5
N3—C5—H9	109.5		
C3—N2—C2—O1	4.0 (2)	N2—C3—C4—O2	129.75 (17)
C3—N2—C2—C1	-176.27 (13)	N2—C3—C4—N3	-52.78 (19)
N1—C1—C2—O1	-45.43 (19)	C4—N3—C5—C6	147.06 (16)
C7—C1—C2—O1	72.41 (19)	N3—C5—C6—O4	7.0(2)
N1—C1—C2—N2	134.83 (14)	N3—C5—C6—O3	-174.41 (14)
C7—C1—C2—N2	-107.33 (16)	N1—C1—C7—C8	-175.12 (14)
C2—N2—C3—C4	-87.54 (18)	C2—C1—C7—C8	68.14 (19)
C5—N3—C4—O2	7.5 (2)	C1—C7—C8—C10	62.1 (2)
C5—N3—C4—C3	-169.93 (15)	C1—C7—C8—C9	-175.13 (17)

### Hydrogen-bond geometry (Å, °)

<i>D</i> —H··· <i>A</i>	<i>D</i> —Н	$H\cdots A$	D··· $A$	<i>D</i> —H··· <i>A</i>
N1—H1···O3 <sup>i</sup>	0.97(3)	1.78 (3)	2.743 (2)	168 (2)
N1—H2···O4 <sup>ii</sup>	0.92(2)	1.94(2)	2.822(2)	160 (2)
N1—H2···O3 <sup>ii</sup>	0.92(2)	2.52(2)	3.260(2)	138 (2)
N1—H3···O1 <sup>iii</sup>	0.89(2)	2.45 (2)	3.031(2)	123 (2)
N1—H3···O3 <sup>iv</sup>	0.89(2)	2.05(2)	2.870(2)	151 (2)
N2—H5···O2 <sup>v</sup>	0.83(2)	2.05 (2)	2.832(2)	155 (2)
C1—H4···O1 <sup>v</sup>	1.00	2.33	3.269 (2)	155

 $\text{Symmetry codes: (i)} \ -x + 1/2, \ -y + 1, \ z - 1/2; \ (ii) \ -x - 1/2, \ -y + 1, \ z - 1/2; \ (iii) \ x - 1/2, \ -y + 3/2, \ -z + 1; \ (iv) \ -x, \ y + 1/2, \ -z + 3/2; \ (v) \ x - 1, \ y, \ z.$